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Key indicators

Single-crystal X-ray study T = 295 KMean $\sigma(C-C) = 0.002 \text{ Å}$ R factor = 0.042 wR factor = 0.125 Data-to-parameter ratio = 14.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. Isatin 3-benzoylhydrazone: a monoclinic form

The amine N atom of the isatin portion of the title compound (systematic name: 2-oxo-2,3-dihydro-1*H*-indole-3-carbalde-hyde benzoylhydrazone), $C_{15}H_{11}N_3O_2$, forms a hydrogen bond with the amide O atom of the benzoylhydrazone portion of a symmetry-related molecule, giving rise to a hydrogen-bonded chain structure that propagates by a glide plane along the *c* axis of the monoclinic unit cell.

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Comment

There are a large number of structurally authenticated Schiff bases that are the condensation products of an aroylhydrazine and a carbonyl compound, as noted from the Cambridge Structural Database (Version 5.26; Allen, 2002). Among this number is a sole example of an isatin derivative, the 3-pico-linoylhydrazone. Isatin aroylhydrazones are expected to function as a terdentate chelates to metals; however, in nickel bis(1-methylisatin 3-picolinoylhydrazonate), the pyridyl N atom coordinates in place of the amide O atom (Rodriguez-Argüelles *et al.*, 2004).



Isatin 3-benzoylhydrazone, (I) (Fig. 1), is a nearly planar molecule. The 3-picolinovlhydrazone mentioned above is also planar, and the planarity of the two compounds can be attributed to delocalization of the π electrons over the entire molecule through the -N-NH-C(O) portion. Other Schiffbase derivatives of isatin that have been crystallographically characterized include (4-hexylphenyl)imino-1H-indol-2(3H-3 one) (Öztürk et al., 2003), indole-2,3-dione 3-(trimethylammonioacetyl)hydrazone chloride (Sun et al., 2001), indole-2,3-dione 3-(hexamethyleneiminyl)thiosemicarbazone (Bain et al., 1997), 2,3-bis(thiosemicarbazone) (Casas et al., 2000) and indole-2,3-dione β -4-(4-tolyl)thiocarbazone (Revenko et al., 1994). Useful biological properties (Öztürk et al., 2003) are expected of the title compound and other Schiff bases. The title compound features an intramolecular hydrogen bond; adjacent molecules are linked by another hydrogen bond to form a chain. The following paper (Ali et al., 2005) describes an orthorhombic polymorph of the title compound.

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Experimental

Benzhydrazide (0.50 g, 3.7 mmol) and isatin (0.54 g, 3.7 mol) were refluxed in ethanol (50 ml) for 2 h. The solid that was formed was filtered off and recrystallized from pyridine to give well defined orange-colored blocks.

 $D_r = 1.392 \text{ Mg m}^{-3}$

Cell parameters from 2811

1878 reflections with $I > 2\sigma(I)$

 $w = 1/[\sigma^2(F_0^2) + (0.0678P)^2]$

where $P = (F_0^2 + 2F_c^2)/3$

+ 0.1615P]

 $(\Delta/\sigma)_{\rm max} = 0.001$ $\Delta\rho_{\rm max} = 0.17 \text{ e} \text{ Å}^{-3}$

 $\Delta \rho_{\rm min} = -0.18 \text{ e} \text{ Å}^{-3}$

Mo $K\alpha$ radiation

reflections

 $\theta = 2.4-26.3^{\circ}$ $\mu = 0.10 \text{ mm}^{-1}$

T = 295 (2) K

Block, orange $0.48 \times 0.32 \times 0.26 \text{ mm}$

 $R_{\rm int} = 0.026$

 $\theta_{\max} = 27.1^{\circ}$ $h = -11 \rightarrow 10$

 $k = -15 \rightarrow 15$

 $l = -15 \rightarrow 13$

Crystal data

 $\begin{array}{l} C_{15}H_{11}N_{3}O_{2} \\ M_{r} = 265.27 \\ \text{Monoclinic, } P2_{1}/c \\ a = 8.9279 \ (6) \\ \text{Å} \\ b = 12.1099 \ (9) \\ \text{Å} \\ c = 11.7567 \ (9) \\ \text{Å} \\ \beta = 95.117 \ (1)^{\circ} \\ V = 1266.0 \ (2) \\ \text{Å}^{3} \\ Z = 4 \end{array}$

Data collection

Bruker SMART area-detector diffractometer φ and ω scans Absorption correction: none 7680 measured reflections 2762 independent reflections

Refinement

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Refinement on F^2

R[F^2 > 2\sigma(F^2)] = 0.042

wR(F^2) = 0.125

S = 1.01

2762 reflections

189 parameters

H atoms treated by a mixture of

independent and constrained

refinement
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Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$\frac{N1 - H1n \cdots O2}{N3 - H3n \cdots O1^{i}}$	0.86 (1)	2.00 (1)	2.712 (2)	139 (2)
	0.87 (1)	2.15 (1)	2.918 (2)	148 (2)

Symmetry code: (i) $x, -y + \frac{3}{2}, z - \frac{1}{2}$.

The C-bound H atoms were placed at calculated positions (C–H = 0.93 Å), and they were included in the refinement in the riding-model approximation with $U_{iso}(H)$ set to $1.2U_{eq}(C)$. The N-bound H atoms were located in a difference Fourier map and refined with a distance restraint of N–H = 0.86 (1) Å.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97*.



Figure 1

ORTEPII (Johnson, 1976) plot of $C_{15}H_{11}N_3O_2$. Displacement ellipsoids are drawn at the 50% probability level, and H atoms are shown as spheres of arbitrary radii. Hydrogen bonds are shown as dashed lines.

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References

- Ali, M. H., Abdul Halim, S. N. & Ng, S. W. (2005). Acta Cryst. E61, o3287– o3288.
- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Bain, G. A., West, D. X., Krejci, J., Valdés-Martinez, J., Hernández-Ortega, S. & Toscano, R. A. (1997). *Polyhedron*, 16, 855–862.
- Bruker (2000). SAINT and SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
- Casas, J. S., Castiñeiras, A., Rodrìguez-Argüelles, M. C., Sánchez, A., Sordo, J., Vázquez-López, A. & Vázquez-López, E. M. (2000). J. Chem. Soc. Dalton Trans. pp. 4056–4063.
- Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.

Öztürk, S., Akkurt, M., Özgür, M. Ü., Erçag, A. & Heinemann, F. W. (2003). Acta Cryst. E**59**, 0569–0571.

Revenko, M. D., Kravtsov, V. K. & Simonov, Yu. A. (1994). Kristallografiya, 39, 50–54.

Rodrìguez-Argüelles, M. C., Ferrari, M. B., Gisceglie, F., Pelizzi, C., Pelosi, G., Pinelli, S. & Sassi, M. (2004). J. Inorg. Biochem. 98, 313–321.

- Sheldrick, G. M. (1997). *SHELXS*97 and *SHELXL*97. University of Göttingen, Germany.
- Sun, Y.-Q., Yang, R.-S., Zhang, H.-H., Sun, R.-Q. & Yang, Q.-S. (2001). Chin. J. Struct. Chem. 20, 310–312.